



>>Entry Name: Aluminium(III) iodide (AlI₃)

Synonym(s): Aluminium triiodide

CAS Registry Number: 7784-23-8

AlI₃

Molecular Formula: AlI₃

Molecular Weight: 407.695

Accurate Mass: 407.694958

Percentage Composition: Al 6.62%; I 93.38%

General Statement: In cryst., AlI₄ tetrahedra share corners to give chains as in SO₃

Use/Importance: Lewis acid. Friedel-Crafts catalyst. Commercially available

Physical Description: White moisture sensitive cryst.

Melting Point: Mp 179-189°

Other Data: ΔH_f° AlI₃(s) -280.4 kJ mol⁻¹, S_{298}° 219.9 J K⁻¹ mol⁻¹. Many complexes descr.

Hazard & Toxicity: Reacts violently with H₂O

Aldrich: 40932-4

>>Derivative: Tris-Py complex

Molecular Formula: C₁₅H₁₅AlI₃N₃

Molecular Weight: 644.999

Accurate Mass: 644.821555

Percentage Composition: C 27.93%; H 2.34%; Al 4.18%; I 59.03%; N 6.51%

Physical Description: White moisture-sensitive solid

Melting Point: Mp 195-198 dec.°

Solubility: Sol. MeOH; spar. sol. most org. solvs.

References:

Inorg. Synth., 1953, **4**, 117; 1980, **20**, 82, (*synth, complexes*)

Segel, S.L., *J. Chem. Phys.*, 1956, **25**, 578, (*nqr*)

O'Reilly, D.E., *J. Chem. Phys.*, 1960, **32**, 1007, (*Al-27 nmr*)

Greenwood, N.N. *et al.*, *Friedel-Crafts Relat. React.*, (Ed. Olah, G.A.), Wiley, N.Y., 1963, **1**, 569, (*uses*)

Beattie, I.R. *et al.*, *J.C.S.(A)*, 1968, 813; 1969, 2655, (*ir, Raman*)

O'Brien, J.F. *et al.*, *J. Phys. Chem.*, 1970, **74**, 743, (*complexes*)

McGuire, G.E. *et al.*, *Inorg. Chem.*, 1973, **12**, 2450, (*pe*)

Anthoney, M.E. *et al.*, *J.C.S. Dalton*, 1973, 659, (*thermodyn*)

Kniep, R. *et al.*, *Angew. Chem., Int. Ed.*, 1982, **21**, 386

Dalibart, M. *et al.*, *Coord. Chem. Rev.*, 1986, **74**, 1, (*complexes*)