



>>Entry Name: Barium metaphosphate, 9Cl, 8Cl

CAS Registry Number: 13762-83-9

Extra CAS Registry Numbers(s): 13847-18-2

[Ba(PO₃)₂]_x

Molecular Formula: BaO₆P₂

Molecular Weight: 295.271

Accurate Mass: 295.822246

Percentage Composition: Ba 46.51%; O 32.51%; P 20.98%

General Statement: Known in amorphous, cryst. and vitreous forms

Source/Synthesis: Synth. from BaCO₃ + (NH₄)₂HPO₄ at 465-500°; by heating Ba(H₂PO₄)₂ at >500°. Both methods yield amorphous salt which cryst. at >=600°

Use/Importance: Component of optical glasses

Melting Point: Mp 875°

Aldrich: 43391-8

Fluka: 11789

Sigma: B2917

>>Variant: α-form

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General Statement: The existence of this phase has been questioned

Source/Synthesis: Stated to be formed from β-form at 785°, and from BaCO₃ + Na₂CO₃ + H₃PO₄ at 380°

Physical Description: Slender plates or lozenge-shaped cryst.

>>Variant: β-form

Molecular Formula: BaO₆P₂

Molecular Weight: 295.271

Accurate Mass: 295.822246

Percentage Composition: Ba 46.51%; O 32.51%; P 20.98%

General Statement: Orthorhombic. P-O 143-165 pm and OPO 95-122°

Source/Synthesis: Synth. from BaCO₃ + H₃PO₄ at 350°

Physical Description: Colourless prisms with pyramidal tetragonal ends

Other Data: At 785°, reversibly yields the α-form

>>Variant: γ-form

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Accurate Mass: 295.822246

Percentage Composition: Ba 46.51%; O 32.51%; P 20.98%

General Statement: Monoclinic. Chains formed of 4 PO₄ tetrahedra sharing corners with P-O 147-163 pm, OPO 97-120°. Ba²⁽⁺⁾ ions are octa coordinated by oxygens in an irregular dodecahedron with Ba-O 269-306 pm

Source/Synthesis: Synth. by the rapid dehydration of Ba(H₂PO₄)₂ at 500-600°

References:

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