>>**Entry Name: Zinc chlorate ($Zn(ClO_3)_2$)**
**Synonym(s):** Zinc chlorate

**CAS Registry Number:** 10361-95-2
$Zn(ClO_3)_2$

**Molecular Formula:** $Cl_2O_6Zn$
**Molecular Weight:** 232.292
**Accurate Mass:** 229.836339
**Percentage Composition:** Cl 30.52%; O 41.33%; Zn 28.15%
**Source/Synthesis:** Synth. from $ZnSO_4$ and $Ba(ClO_3)_2$
**Use/Importance:** Bleaching agent; component of explosives; herbicide
**Physical Description:** Hygroscopic cryst.
**Solubility:** V. sol. $H_2O$ (262g per 100g at 20°), EtOH; sol. $Me_2CO$
**Other Data:** Various hydrates known which can be dehydrated to the anhyd. salt
**Hazard & Toxicity:** Powerful oxidant

>>**Derivative:** Tetrahydrate

**Molecular Formula:** $Cl_2H_8O_{10}Zn$
**Molecular Weight:** 304.353
**Accurate Mass:** 301.878599
**Percentage Composition:** Cl 23.30%; H 2.65%; O 52.57%; Zn 21.48%
**Physical Description:** Yellowish to colourless deliquescent cubes
**Solubility:** V. sol. $H_2O$, EtOH, $Me_2CO$, $Et_2O$, glycerine

**References:**


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