



>>Entry Name: Tin

Synonym(s): Stannum (L.)

CAS Registry Number: 7440-31-5

Related CAS Registry Numbers(s): 12597-39-6 15700-33-1 22537-50-4 22541-90-8 26288-30-2 53237-30-2
Sn

Molecular Formula: Sn

Molecular Weight: 118.71

Accurate Mass: 117.9022

Percentage Composition: Sn 100.00%

Essential trace element. Abundance: 2.1 ppm (earth's crust); 3×10^{-3} ppm (sea water)

Source/Synthesis: Cassiterite (SnO_2) is the only important ore, but other, sulfide ores e.g. stannite, sometimes worked. Oxide is smelted, the impure metal refined electrolytically

Use/Importance: Important as coatings and alloys, e.g. Bronzes (with Cu), *Pewter* (with Sb), and type metals (with Pb, Sb, Bi). Radioactive ^{110}Sn used as a tracer

Melting Point: Mp 232°

Boiling Point: Bp 2625°

Aldrich: 24343-4

Fluka: 96523

Sigma: T1036

>>Variant: α -form

Synonym(s): Grey tin

Molecular Formula: Sn

Molecular Weight: 118.71

Accurate Mass: 117.9022

Percentage Composition: Sn 100.00%

Physical Description: Reacts with conc. HCl (aq.) $\rightarrow \text{SnCl}_4$

Other Data: $d^{20} 5.77 \text{ g cm}^{-3}$. Transforms into the β -form at 13.2°

>>Variant: β -form

Synonym(s): White tin

Molecular Formula: Sn

Molecular Weight: 118.71

Accurate Mass: 117.9022

Percentage Composition: Sn 100.00%

Physical Description: Soft, pliable, silvery-white metal. Reacts with conc. HCl (aq.) $\rightarrow \text{SnCl}_2$. Unreactive to O_2 (film protection). Dissolves in acids and bases with liberation of H_2 . Steam $\rightarrow \text{H}_2$. Reacts with Cl_2 , Br_2 in cold and F_2 , I_2 on warming. Reacts vigorously with S, Se when heated

Other Data: $d^{20} 7.27 \text{ g cm}^{-3}$

>>Variant: Tin 108

CAS Registry Number: 24558-38-1

Source/Synthesis: Prod. from Cd ($\alpha, 2n$)

Other Data: $I = +0$. Decays by orbital electron capture or $\beta^{(+)}$ emission to ^{108}In

>>Variant: Tin 111

CAS Registry Number: 15720-78-2

Source/Synthesis: Prod. from Cd ($\alpha, 3n$), or ^{112}Sn (γ, n)

Other Data: $I = +7/2$; $\mu = +0.61 \mu_N$. Decays by orbital electron capture or $\beta^{(+)}$ emission to ^{111m}In

>>Variant: Tin 112

CAS Registry Number: 15125-53-8

Other Data: $I = +0$