>>**Entry Name: Sodium selenite (Na_2SeO_3)**
Synonym(s): Disodium selenate(*IV*). Selenious acid disodium salt, 8Cl. Selenic(*IV*) acid disodium salt

CAS Registry Number: 10102-18-8
Extra CAS Registry Numbers(s): 14013-56-0
Na_2SeO_3

Molecular Formula: Na_2O_3Se
Molecular Weight: 172.938
Accurate Mass: 173.880799
Percentage Composition: Na 26.59%; O 27.75%; Se 45.66%
Source/Synthesis: Synth. by reaction of Na_2SeO_4 with Na_2CO_3 at 700-800°. Synth. from mechanochemical activation (vibrational mill) of SeO_2/Na_2CO_3 mixt.
Use/Importance: Commercially available. Used in the electroplating of copper and in the prep. of electrophotographic photoreceptors
Physical Description: Moisture sensitive white powder. Undergoes monoclinic to hexagonal transformation at 617°. Dec. on heating to form Na_2O and SeO_2 which dec. further to Se and O_2
Melting Point: Mp 710°
Solubility: V. sol. H_2O (85 g per 100 cm^3 at 20°)
Other Data: ΔH_f° –958 kJ mol^{-1}
Hazard & Toxicity: Highly toxic

Aldrich: 21448-5
Fluka: 71950
Sigma: S5779

>>**Derivative:** Pentahydrate

CAS Registry Number: 26970-82-1
Molecular Formula: $H_{10}Na_2O_8Se$
Molecular Weight: 263.014
Percentage Composition: H 3.83%; Na 17.48%; O 48.66%; Se 30.02%
Source/Synthesis: Commercially available. Cryst. from Na_2SeO_3 aq. at <35°
Physical Description: White tetragonal cryst.; dec. on heating by H_2O loss
Solubility: Sol. H_2O; insol. EtOH
Hazard & Toxicity: Highly toxic

Aldrich: 30845-5
Fluka: 71955

>>**Derivative:** Octahydrate

CAS Registry Number: 41292-05-1
Molecular Formula: $H_{16}Na_2O_{11}Se$
Molecular Weight: 317.059
Percentage Composition: H 5.09%; Na 14.50%; O 55.51%; Se 24.90%
Source/Synthesis: Cryst. from aq. soln. at <–8°
Physical Description: Colourless cryst.
Hazard & Toxicity: Highly toxic

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