>>**Entry Name: Sodium sulfite (Na_2SO_3)**
ENTRY UNDER REVIEW
Synonym(s): Disodium sulfite. Sulfurous acid disodium salt, 8Cl

CAS Registry Number: 7757-83-7
Extra CAS Registry Numbers(s): 10102-15-5 10579-83-6

Na_2SO_3

Molecular Formula: Na_2O_3S
Molecular Weight: 126.044
Accurate Mass: 125.936349
Percentage Composition: Na 36.48%; O 38.08%; S 25.44%
Source/Synthesis: Synth. by reaction of Na_2CO_3 or NaOH with SO_2 gas. Alternatively obt. by treatment of NaCl with $(NH_4)_2SO_3$. Cryst. from aq. soln. as the heptahydrate which dehydrates when heated to 200°
Use/Importance: Commercially available. Antioxidant. Used in photographic processing and as a pulping agent (normally with NaOH or $NaHCO_3$) for wood, waste paper, etc.
Physical Description: Moisture sensitive white hexagonal cryst.
Solubility: V. sol. H_2O (13 g per 100 cm^3 at 0°; 28 g per 100 cm^3 at 80°); spar. sol. EtOH
Other Data: ΔH_f° –1090 kJ mol^{-1}; ΔG_f° –1002 kJ mol^{-1}; S° 146 J K^{-1} mol^{-1}
Hazard & Toxicity: Irritant

Aldrich: 23932-1
Fluka: 71988
Sigma: S4672

References:

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