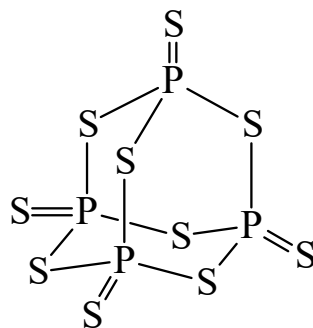




>>Entry Name: Phosphorus sulfide (P₄S₁₀)

Synonym(s): 2,4,6,8,9,10-Hexathia-1,3,5,7-tetraphosphatricyclo[3.3.1.1^{3,7}]decane 1,3,5,7-tetrasulfide, 9Cl. 2,4,6,8,9,10-Hexathia-1,3,5,7-tetraphosphaadamantane 1,3,5,7-tetrasulfide. Phosphorus pentasulfide (incorr.). Tetraphosphorus decasulfide



CAS Registry Number: 1314-80-3

Extra CAS Registry Numbers(s): 15857-57-5

Molecular Formula: P₄S₁₀

Molecular Weight: 444.555

Accurate Mass: 443.615748

Percentage Composition: P 27.87%; S 72.13%

Source/Synthesis: Synth. from P + S at 350-400°

Use/Importance: Thiation reagent, converting eg. amides → thioamides. Deoxygenating agent, eg. sulfoxides → sulfides. Intermed. in manuf. of insecticides, lubricant additives, mineral flotation and extraction agents

Physical Description: Triclinic light yellow, almost colourless cryst. Dec. in moist air, or H₂O. Dissoc. at beginning of vaporisation → S + P₄S₇ and P₄S₃

Melting Point: Mp 281-283° (276°)

Boiling Point: Bp 514°

Solubility: Sl. sol. CS₂ (0.2g per 100g)

Other Data: ΔH₂₉₈^o -680 kJ mol⁻¹

Aldrich: 23210-6

Fluka: 79600

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