



>>Entry Name: Sodium chloride (NaCl), 8Cl, JAN, USAN

Synonym(s): Rock salt. Salt. Halite. Ayr. Hypersal. Brine. Natritope chloride

CAS Registry Number: 7647-14-5

Related CAS Registry Numbers(s): 17112-21-9

Extra CAS Registry Numbers(s): 14762-51-7

NaCl

Molecular Formula: ClNa

Molecular Weight: 58.442

Accurate Mass: 57.958619

Percentage Composition: Cl 60.66%; Na 39.34%

Source/Synthesis: Mineral. Manufactured from salt deposits by conventional extraction methods

Use/Importance: Commercially available. Pharmaceutic aid (tonicity agent). Tablet and capsule diluent. Used in fluxes, electroplating and as a food additive. Raw material for the production of many commercially significant sodium salts. Also used in the preparation of sodium metal

Physical Description: Colourless cubic cryst., hygroscopic

Melting Point: Mp 801°

Boiling Point: Bp 1413°

Solubility: V. sol H<sub>2</sub>O (36 g per 100 cm<sup>3</sup> at 0°; 39 g per 100 cm<sup>3</sup> at 100°); spar. sol. EtOH

Other Data: <sup>22</sup>Na labelled compd. used as radioactive agent (Sodium chloride Na 22, USAN). Component of Lytears.  $\Delta H_f^\circ -411.1 \text{ kJ mol}^{-1}$ ;  $\Delta G_f^\circ -384.0 \text{ kJ mol}^{-1}$ ;  $S^\circ 72.1 \text{ J K}^{-1} \text{ mol}^{-1}$

**Hazard & Toxicity:** Adverse therapeutic effects include hypertension, oedema, and gastrointestinal disturbances. Excessive administration can lead to hypernatraemia with resultant dehydration of internal organs

Aldrich: 31879-5

Fluka: 24889

Sigma: W3504

#### References:

Nickels, J.E. *et al.*, *J. Phys., Colloq.*, 1949, **53**, 625

Kaufmann, D.W., *Sodium Chloride*, ACS Monograph Ser. 145, Reinhold, NY, 1960, (rev)

*Gmelin Handbook Inorg. Chem., Syst. No. 21*, 1964, **1**, 126, (rev, bibl)

Krishnamurthy, N. *et al.*, *Z. Phys.*, 1965, **183**, 130, (Raman)

Clough, R.W., *Handb. Ind. Chem.*, 1974, 123, (rev, uses)

Lowenheim, F.A. *et al.*, *Industrial Chemicals*, (Eds.), 4th Ed., Wiley-Interscience, N.Y., 1975, 722, (manuf)

Heiss, J.F. *et al.*, *Kirk-Othmer Encycl. Chem. Technol.*, 3rd edn., Wiley, 1978, **21**, 205, (rev)

Kare, M.R. *et al.*, *Biol. Behav. Aspects Salt Intake, [Int. Conf.]*, 1979, Academic Press, New York, 1980, (book)

Chase, M.W. *et al.*, *J. Phys. Chem. Ref. Data*, 1985, **14**, 770, (thermochem data)

Tabeta, R. *et al.*, *Bull. Chem. Soc. Jpn.*, 1986, 1986, **59**, 1957, (Na-23 nmr)

*Sigma-Aldrich Library of Chemical Safety Data*, 1988, **2**, 3135d, (haz)

*Handbook of Pharmaceutical Excipients*, 2nd edn., (eds. Wade, A. *et al.*), American Pharmaceutical Association/Pharmaceutical Press, 1994, 439-442

*Martindale, The Extra Pharmacopoeia*, 31st edn., Pharmaceutical Press, 1996, 1184

Lewis, R.J., *Sax's Dangerous Properties of Industrial Materials*, 8th edn., Van Nostrand Reinhold, 1992, SFT000