



>> **Name: Ammonium magnesium phosphate (1:1:1)**, 9Cl, 8Cl
Synonym(s): Phosphoric acid, ammonium magnesium salt (1:1:1), 8Cl

CAS Registry Number: 7785-21-9
Related CAS Registry Numbers(s): 19004-55-8
Mg(NH4)PO4

Molecular Formula: H4MgNO4P
Molecular Weight: 137.315
Accurate Mass: 136.972838
Percentage Composition: H 2.94%; Mg 17.70%; N 10.20%; O 46.61%; P 22.56%

Source/Synthesis: Synth. from Mg(OH)2-H3PO4-NH4OH and Na3PO4-MgSO4-NH4OH. Also MgHPO4 + NH3. Occurs in human urinary calculi, and also guano
Use/Importance: Precipitated during course of gravimetric estimation of Mg. Ignited to Mg2P2O7. Agricultural fertiliser
Physical Description: Amorph. powder
Other Data: A trihydrate has been reported

Fluka: 9884

>>**Derivative:** Monohydrate

CAS Registry Number: 16674-60-5
Source/Synthesis: Obt. from hexahydrate at 80-100°. May be cryst. from aq. solns. containing MgCl2 and [NH4]H2PO4
Solubility: Sol. acids; prac. insol. H2O (0.023 g per 100 cm³ at 0°), (0.02 g per 100 cm at 80°); insol. EtOH
Other Data: Stable below 130-150°

>>**Derivative:** Hexahydrate

CAS Registry Number: 13478-16-5
Physical Description: White powder or elongated platelets
Other Data: Orthorhombic with regular PO4^{3(-)} tetrahedra and distorted Mg(H2O)6^{2(+)} octahedra. A slurry in H2O at 95° → the monohydrate. When heated dry, loses H2O and NH3 in three overlapping processes

>>**Variant:** Mineral-form
Synonym(s): Struvite. Guanite

CAS Registry Number: 15490-91-2
Molecular Formula: H4MgNO4P
Molecular Weight: 137.315
Accurate Mass: 136.972838
Percentage Composition: H 2.94%; Mg 17.70%; N 10.20%; O 46.61%; P 22.56%

Source/Synthesis: Found in human urinary sediments; peat beds; phosphate caves; diatomites, and bird and bat guano deposits in Chile and Australia. Associated with Newberyite and Brushite

Physical Description: Colourless to pale yellowish-brown, transparent to translucent, brittle, bipyramidal cryst.

References:

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Rothbaum, H.P. *et al.*, *N.Z. J. Sci.*, 1973, **16**, 539, (synth, props)
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Paulik, F. *et al.*, *J. Therm. Anal.*, 1975, **8**, 557; 567, (thermal dec, hexahydrate)
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Abdelrazig, B.E.I. *et al.*, *Thermochim. Acta*, 1988, **129**, 197, (props)