



>>Entry Name: Neodymium nitrate

Synonym(s): Trinitratoneodymium

CAS Registry Number: 10045-95-1

Related CAS Registry Numbers(s): 14517-29-4 16454-60-7 26635-06-3

$\text{Nd}(\text{NO}_3)_3$

Molecular Formula: N_3NdO_9

Molecular Weight: 330.255

Accurate Mass: 327.871176

Percentage Composition: N 12.72%; Nd 43.68%; O 43.60%

Physical Description: Hygroscopic violet solid

Solubility: Sol. H_2O , EtOH

Other Data: Hexahydrate normally obt. on cryst; tetra- and pentahydrates also reported

>>Derivative: Hydrate

CAS Registry Number: 13746-96-8

Other Data: Contains variable amts. H_2O

Aldrich: 21720-4

>>Derivative: Hexahydrate

Synonym(s): Neodymium nitrate hexahydrate

CAS Registry Number: 63643-73-2

Molecular Formula: $\text{H}_{12}\text{N}_3\text{NdO}_{15}$

Molecular Weight: 438.346

Accurate Mass: 435.934566

Percentage Composition: H 2.76%; N 9.59%; Nd 32.91%; O 54.75%

Use/Importance: Commercially available

Physical Description: Violet cryst.

Solubility: Sol. H_2O , EtOH, Me_2CO

Other Data: Contains 10-coord. $\text{Nd}(\text{NO}_3)_3(\text{H}_2\text{O})_4$ molecules; isomorphous with Pr analogue

>>Derivative: DMF complex, trihydrate

Synonym(s): Triaqua(*N,N*-dimethylformamide)trinitratoneodymium

Molecular Formula: $\text{C}_3\text{H}_{13}\text{N}_4\text{NdO}_{13}$

Molecular Weight: 457.395

Accurate Mass: 454.955635

Percentage Composition: C 7.88%; H 2.86%; N 12.25%; Nd 31.54%; O 45.47%

Physical Description: Rose cryst. + H_2O

Other Data: Has 10-coord. Nd (nitrates bidentate) in approx. bicapped square antiprismatic geometry

>>Derivative: Bis- Ph_3PO complex

CAS Registry Number: 14706-33-3

Molecular Formula: $\text{C}_{36}\text{H}_{30}\text{N}_3\text{NdO}_{11}\text{P}_2$

Molecular Weight: 886.834

Accurate Mass: 884.04328

Percentage Composition: C 48.76%; H 3.41%; N 4.74%; Nd 16.26%; O 19.85%; P 6.99%

Physical Description: Pale lilac needles + EtOH (EtOH/ CHCl_3)

Melting Point: Mp 161-170 dec.° (EtOH adduct)

Solubility: Sl. sol. EtOH

Other Data: 9-Coord. with bidentate nitrates and EtOH coord.; EtOH cannot be removed on heating

References: