>>**Entry Name: Lead(*II*) oxide (PbO),** 9CI
Synonym(s): Lead monoxide

CAS Registry Number: 1317-36-8
PbO

Molecular Formula: OPb
Molecular Weight: 223.199
Accurate Mass: 223.971542
Percentage Composition: O 7.17%; Pb 92.83%
Source/Synthesis: Pure compd. is synth. from $Pb(OAc)_2$ and NH_4OH
Use/Importance: Used for plates in electric storage batteries of the lead sulfuric acid type. Widely used in optical, electrical, and electronic glasses
Melting Point: Mp 897°
Solubility: Insol. H_2O, EtOH; sol. AcOH
Other Data: Other lead oxides are listed under Lead oxide (Pb_2O_3) KBS41-J, Lead oxide (Pb_3O_4) KBS14-D and Lead(*IV*) oxide (PbO_2) KBS44-M
Hazard & Toxicity: Toxic. LD_{Lo} (rat, ipr) 430 mg/kg

Aldrich: 20361-0
Fluka: 15338
Sigma: L4281
Other: Gelest PBL6380

>>**Variant:** Tetragonal-form
Synonym(s): Litharge

Molecular Formula: OPb
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opposite
Source/Synthesis: Prepd. by treating PbO_2 in air to 550°. Also mineral from e.g. California, Greece
Physical Description: Red to reddish-yellow tetragonal cryst.

>>**Variant:** Orthorhombic-form
Synonym(s): Massicot

Molecular Formula: OPb
Molecular Weight: 223.199
Accurate Mass: 223.971542
Percentage Composition: O 7.17%; Pb 92.83%

Source/Synthesis: Mineral: oxidation product of primary lead minerals esp. in association with galena
Physical Description: Yellow cryst., stable from above 489°; metastable at r.t. Amphoteric

References:

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