>>**Entry Name: Lithium ethoxide**
Synonym(s): Ethanol lithium salt, 9CI. Ethyl alcohol lithium salt, 8CI. Ethoxylithium

CAS Registry Number: 2388-07-0
LiOEt

Molecular Formula: C_2H_5LiO
Molecular Weight: 52.002
Accurate Mass: 52.050043
Percentage Composition: C 46.19%; H 9.69%; Li 13.35%; O 30.77%
Source/Synthesis: Prepd. by dissolution of Li metal in excess EtOH; isol. and purif. by vacuum sublimation
Use/Importance: Component of polym. and dehydrochlorination catalysts. Nucleophilic substitution reagent. Lithiation reagent. Forms complexes with EtLi in hydrocarbon solvs.
Physical Description: Colourless powder. Moisture sensitive. Reacts with H_2O to form LiOH and EtOH. Reacts with acidic protons to form the corresp. Li salts. Dec. on heating in a complex reaction giving H_2 and C_2H_4
Solubility: Sol. EtOH (12.5 g per 100 cm^3 at 24°, 18.2 g per 100 cm^3 at 70°); spar. sol. heptane (0.26 g per 100 cm^3 at 26°), pentane, Et_2O
Other Data: ΔH_f° –532 kJ mol^{-1}

Aldrich: 40020-3
Fluka: 62494
Other: Gelest AKL458

References:

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