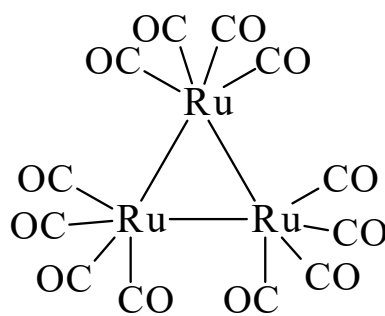




>>Entry Name: Dodecacarbonyltriruthenium, 9Cl

Synonym(s): Triruthenium dodecacarbonyl. Ruthenium carbonyl ($\text{Ru}_3(\text{CO})_{12}$)



CAS Registry Number: 15243-33-1

Type of Compound Code(s): BA0650 BA0180 BA0610 BA0300 BA0490 BA0040 BA0270 BA0070 BA0095 BA0275 BA0482 BA0495 BA0290 BA0440 BA0407 BA0080 BA0345 BA0230 BA0060 BA0430 BA0130 BA0235 BA0170 BA0320 BA0030 BA0360 BA0460 BA0330

Molecular Formula: $\text{C}_{12}\text{O}_{12}\text{Ru}_3$

Molecular Weight: 639.335

Accurate Mass: 641.652024

Percentage Composition: C 22.54%; O 30.03%; Ru 47.43%

Source/Synthesis: Synth. either by high pressure carbonylation of a wide range of ruthenium compds. (e.g. RuCl_3 , $\text{Ru}(\text{acac})_3$ and Ru^{III} carboxylates) or by treating a 2-ethoxyethanol soln. of RuCl_3 with CO, followed by Zn redn.

Use/Importance: A vastly important reagent. Active catalyst for many processes including: water-gas shift reaction; addn. of carboxylic acids to alkynes $\rightarrow$ vinyl esters; hydrogenation of alkynes, dienes, polyenes; Fischer-Tropsch reaction; decarbonylation of formates $\rightarrow$ alcohols; allylation of RCHO with allyl acetates in presence of CO/ Et_3N $\rightarrow$ homoallyl alcohols; *N*-alkylation of amines by alcohols, diols in presence of PBU_3 ; carbonylation of cyclic amines $\rightarrow$ formamides; transfer hydrogenolysis of CX_4 ($\text{X} = \text{Cl}, \text{Br}$); hydroformylation and hydroacylation of alkenes; carbonylation/homologation of alcohols; homologation of HOAc with CO; carbonylation of $\text{ArNO}_2 \rightarrow$ isocyanates; precursor to catalysts for rearrangement of azobenzenes $\rightarrow$ 1-phenyl-2-alkylbenzimidazoles; isom. of alkenes, dienes; transfer hydrogenation of ketones with MeOH; dehydrogenation of alcohols $\rightarrow$ esters; oxidn. of alkynyl ethers, amines by PhIO $\rightarrow$ α -keto esters, α -keto amides; redn. of ArNO_2 to ArNH_2 using CO/ H_2O or CO/ H_2 ; reaction of alkenes with CO in presence of base $\rightarrow$ ketones; hydrosilylation of fluorinated alkenes; with $\text{Fe}_3(\text{CO})_{12}$, reductive amination of ketones; amidocarboxylation of alkynes $\rightarrow$ enol carbamates; reductive *N*-heterocyclisation; reductive carbonylation of ArNO_2 with CO/MeOH $\rightarrow$ ArNHCOOMe ; H/D exchange in tertiary amines; alcohol oxidn. with Bu^tOOH ; dehydrogenative hydrosilylation of alkenes/r vinyl silanes; reductive decarboxylation of acids; hydrocarbonylation of ethyne $\rightarrow$ hydroquinone; *ortho*-acylation of pyridines with $\text{RCH}=\text{CH}_2/\text{CO}$; silylcarbonylation of alkenes $\rightarrow$ silyl enol ethers; amines/ $\text{CO}_2 \rightarrow$ ureas; $\text{RCHO} + \text{R}'\text{CH}_2\text{OH} \rightarrow \text{RC}(\text{O})\text{OCH}_2\text{R}'$; Pauson-Khand cocyclisation of alkynes, alkenes and CO; undergoes subst. reactions; can be converted to other Ru cluster complexes

Physical Description: Air-stable orange cryst. (hexane or C_6H_6)

Melting Point: Mp 154-155 dec. $^\circ$

Boiling Point: Subl. 80-100 $^\circ$ *in vacuo*

Solubility: Sol. most org. solvs. (e.g. hexane, C_6H_6 , CHCl_3 , EtOH); insol. MeOH, H_2O

Other Data: $\Delta H_f^\circ -1903 \text{ kJ mol}^{-1}$; $\Delta H_{\text{fusion}} 43.5 \text{ kJ mol}^{-1}$

Aldrich: 24501-1

Fluka: 84055

References: