



>>Entry Name: Aluminium tellurides

Molecular Formula: $[\text{AlTe}]$

>>Variant: Aluminium telluride ($\text{Al}_7\text{Te}_{10}$)

CAS Registry Number: 110772-85-5

Molecular Formula: $\text{Al}_7\text{Te}_{10}$

Molecular Weight: 1464.87

Accurate Mass: 1487.933063

Percentage Composition: Al 12.89%; Te 87.11%

Source/Synthesis: Synth. from elements at 927°

Physical Description: Air- and moisture-sensitive ruby red cryst.

>>Variant: Aluminium telluride (Al_2Te_3)

CAS Registry Number: 12043-29-7

Molecular Formula: Al_2Te_3

Molecular Weight: 436.763

Accurate Mass: 443.681765

Percentage Composition: Al 12.36%; Te 87.64%

Source/Synthesis: Synth. from elements at 1000°

Use/Importance: Source of H_2S (D_2S) as *in situ* reducing agent for aldehydes and enones. Commercially available

Physical Description: Dark grey solid (melt)

Melting Point: $\text{Mp } 895^\circ$

Other Data: $\Delta H_f^\circ -326 \text{ kJ mol}^{-1}$. Band gap E_g 2.2 eV. Conductance $1.85 \times 10^4 \Omega^{-1} \text{m}^{-1}$

References:

- Kauer, E. *et al.*, *Z. Naturforsch., A*, 1958, **13**, 531, (*optical props*)
Steigmann, G.A. *et al.*, *Acta Cryst.*, 1966, **20**, 617, (*cryst struct*)
Gromakov, S.D. *et al.*, *Russ. J. Phys. Chem. (Engl. Transl.)*, 1972, **46**, 91, (*props*)
Said, H. *et al.*, *J. Less-Common Met.*, 1976, **46**, 209, (*thermodyn*)
Kambe, N. *et al.*, *Angew. Chem., Int. Ed.*, 1980, **19**, 1009, (*use*)
Nesper, R. *et al.*, *Z. Naturforsch., B*, 1987, **42**, 557, (*synth, cryst struct*)